

Materials of Conferences

**FORMATION OF OXIDE TUNGSTEN
BRONZES ON W – SUBSTRATE
BY ELECTROCHEMICAL SYNTHESIS
FROM MOLTEN SALTS**

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The work is related to the field of high-temperature electrochemistry, in particular, obtaining nanostructural coatings of oxide tungsten bronzes (OTB) by molten salts electrolysis [1–5]. The obtained samples may find application in medicine, electrical engineering, radio engineering, food and chemical industries.

The following melts were used in this study:

1. $0,30 K_2WO_4 - 0,25 Li_2WO_4 - 0,45 WO_3$.
2. $0,25K_2WO_4 - 0,25Na_2WO_4 - 0,50WO_3$.
3. $0,55 Li_2WO_4 - 0,45 WO_3$.

The polycrystalline coatings OTB ($K_xLi_yWO_3$) of hexagonal structure were obtained (Fig. 1) during the electrolysis in the $0,30 K_2WO_4 - 0,25 Li_2WO_4 - 0,45 WO_3$ melt under pulsed potentiostatic conditions on the textured W-plate. Each microcrystal is an oriented nano-needle structure, in which the needle clusters are linked between each other with the intermediate neck. By increasing the pulse duration up to several seconds the space between the nano-needles becomes filled by layer-by-layer growth from the needles along the intermediate neck and formation of hexagonal OTB plates with smooth faces occurs (Fig. 2).

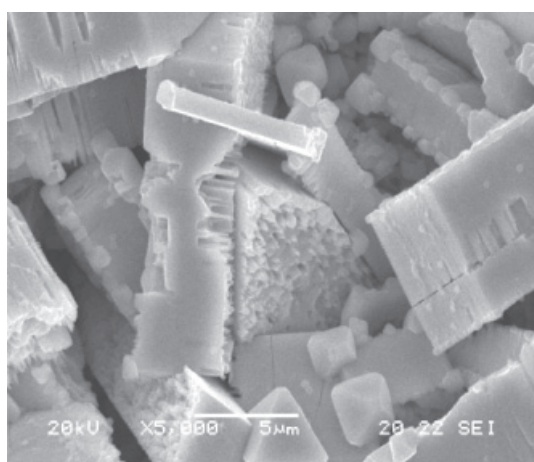


Fig. 1. Morphology of the hexagonal OTB nanocrystalline deposit:
 $T = 700^\circ C$, $\eta = 200$ mV, $t = 0,5$ s

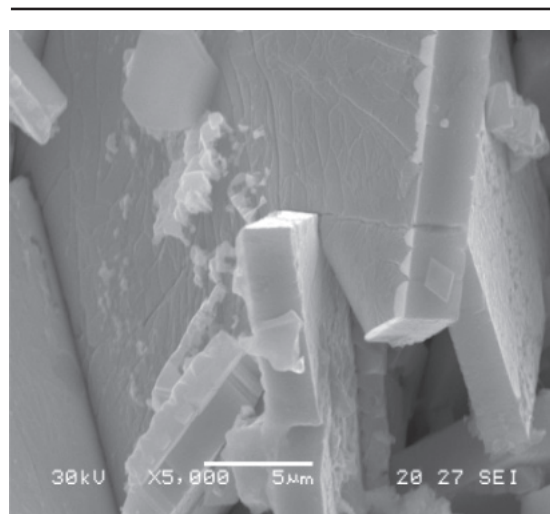


Fig. 2. Morphology of the hexagonal OTB:
 $T = 700^\circ C$, $\eta = 200$ mV, $t = 15$ s

The unformed tetragonal prisms are deposited from the $0,25 K_2WO_4 - 0,25 Na_2WO_4 - 0,50 WO_3$ melt (Fig. 3). The tops of individual crystals are composed of needles, having the orientation $\langle 001 \rangle$ (the needle thickness at the half-height is 80–150 nm). The process of formation of the regular OTB microcrystal ($K_xNa_yWO_3$) with the tetragonal structure was found to be faster than that of the microcrystal with the hexagonal structure. The formation time is 0,5–1 s.

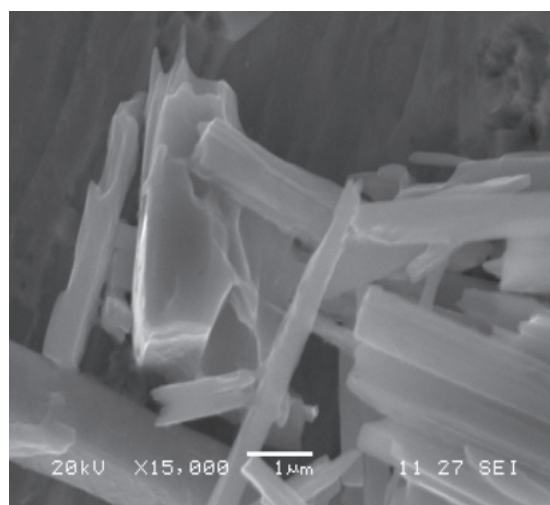


Fig. 3. Morphology of the tetragonal OTB:
 $T = 700^\circ C$, $\eta = 130$ mV; $t = 0,2$ s

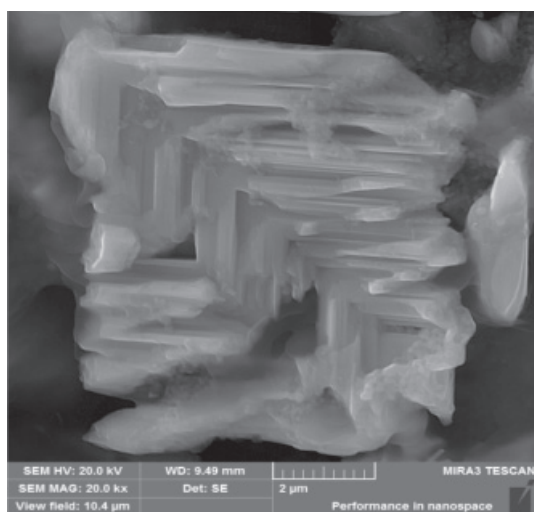


Fig. 4. Microcrystal of the cubic OTB:
 $T = 800^{\circ}\text{C}$, $\eta = 250 \text{ mB}$, $\tau = 0,1 \text{ s}$

During the $0,55 \text{ Li}_2\text{WO}_4 - 0,45 \text{ WO}_3$ melt electrolysis the polycrystalline deposits consisting of cubic OTB crystals (Li_3WO_3) are formed. These crystals are formed by the intergrowth of separate needle clusters. The formation mechanism of OTB with different structures (cubic, tetragonal and hexagonal) is assumed to be the same and includes the following stages: the formation of nanoclusters; their interaction with the formation of the intermediate neck; the subsequent filling of the space between the needles. At the same time, these clusters can be positioned both in parallel, as in the case of

tetragonal and hexagonal OTB, and perpendicularly to each other, forming stepped structures in the case of cubic OTB (Fig. 4). In the process of growth the stepped structures become porous and then form crystals with smooth faces.

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References

1. Vakarín S.V. Patent RF, № 2354753. Published: May 10, 2009.
2. Vakarín S.V., Melyaeva A.A., Pankratov A.A., Kochedykov V.A., Akashev L.A., Plaksin S.V., Zaykov Yu.P. Patent RF, № 2426822. Published: June 20, 2011.
3. Vakarín S.V., Melyaeva A.A., Semerikova O.L., Konratyuk V.S., Pankratov A.A., Plaksin S.V., Porotnikova N.M., Zaykov Yu.P., Petrov L.A., Mikushina Yu.V., Shishmakov A.B., Chupakhin O.N. Patent RF, № 2456079. Published: May 10, 2012.
4. Vakarín S.V., Melyaeva A.A., Semerikova O.L., Konratyuk V.S., Pankratov A.A., Plaksin S.V., Porotnikova N.M., Zaikov J.P., Petrov L.A., Mikushina Y.V., Shishmakov A.B., Chupakhin O.N., Russian Chemical Bulletin, International Edition (Translation of Izvestiya Akademii Nauk, Seriya Khimicheskaya). – 2011. – Vol. 60, № 10. – P. 1985–1988.
5. Vakarín S.V., Semerikova O.L., Surat S.A., Pankratov A.A., Zaikov Yu.P. Tsvetnye Metally (Non-ferrous metals). – 2013. – № 12. – P. 71–76.

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